

Uptimax Ni-Cd battery

Technical manual



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1. Introduction

The nickel-cadmium battery is the most reliable battery system available in the market today. Its unique features enable it to be used in applications and environments untenable for other widely available battery systems. With the advent of the valve-regulated lead acid battery a new concept was available to the customer, a battery that did not require water replenishment. However, this was obtained at the cost of reliability. To give the customer a highly reliable battery of zero or ultra-low maintenance Saft has developed the Uptimax low maintenance pocket plate battery.

This publication details the design and operating characteristics of the Saft Nife brand Uptimax battery to enable a successful battery system to be achieved. A battery which in normal application requires only one topping-up during its entire service life but has all the well-proven advantages of the nickel-cadmium pocket plate battery.

2. Battery applications

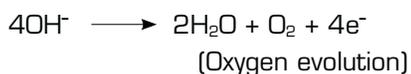
- UPS
- Process control
- Emergency systems
- Security systems
- Offshore oil and gas
- Switchgear

Uptimax batteries are designed to supply the ideal low maintenance power backup solution for installations that demand maximum reliability and optimum TCO (total cost of ownership) while operating for long periods at high ambient temperatures. Uptimax is especially suited for the oil and gas, utility and electricity industries.

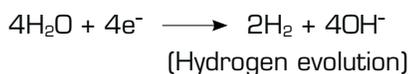
3. Principles of the oxygen recombination cycle

In a conventional flooded electrolyte pocket plate nickel-cadmium battery water is lost from the battery on overcharge due to the following reactions:

At the positive plate



At the negative plate



This corresponds to a theoretical loss of 36 g of water for 107 Ah of overcharge i.e. 0.335 cm³ per Ah. Hence a conventional cell requires periodic addition of water. The frequency of this operation depends upon the cumulative amount of charge received and the operating temperature.

During the charging process evolution of oxygen begins to occur a little before the positive plate reaches its fully charged state and then becomes the main reaction when the fully charged condition is reached. However, the cadmium negative plate has a better charge acceptance than the positive plate and hydrogen is not evolved until this plate is virtually fully charged.

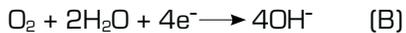
The Uptimax battery has been designed with an excess of cadmium negative material to enhance this effect and ensure that oxygen evolution commences prior to hydrogen evolution.

The oxygen which is produced at the positive plate surface is collected by the special porous separator and thus not allowed to escape from the region between the plates. Some displacement of electrolyte within the separator occurs, thus generating extra unfilled pores for the diffusion of oxygen directly to the adjacent cadmium negative plate.

As soon as the oxygen reaches the negative plate it reacts either chemically:



or electrochemically:



Reaction (A) has the effect of chemically discharging some of the cadmium to cadmium hydroxide. The current passing through the battery is used to recharge this material.

Reaction (B) consumes the current directly. Thus hydrogen evolution at the negative plate is suppressed because the preferred reaction is oxygen recombination. Hence the total process of oxygen generation and consumption is referred to as an oxygen recombination cycle.

The efficiency of this oxygen recombination process depends upon the relationship between the rate at which oxygen is produced and the rate at which it can be collected and transferred to the negative plate surface. The rate of collection and transfer of oxygen is controlled by the separator type and the cell design.

The rate at which oxygen is produced on overcharge is directly related to the charge current once the positive plate has reached a full state of charge. The charge current in turn is controlled by the charging voltage level set on the charging equipment and the ambient temperature. By controlling the charge voltage high efficiencies can be obtained and in this way the rate of water loss can be reduced to a fraction of that from conventional batteries.

Though the efficiency of this oxygen recombination is high it will never achieve 100 % as small quantities of oxygen will escape from the separator before reaching and reacting at the negative plate. Thus a small quantity of hydrogen will ultimately be generated and hence a low rate of water loss will occur. The battery is designed to accommodate this by provision of a generous electrolyte reserve both above and around each cell pack within the battery. This ensures a long service life without the need to replenish with water too often.

4. Construction features of the Uptimax battery

The construction of the Saft Nife brand Uptimax cell is based upon the Saft pocket plate technology but with special features to enhance the low water usage by means of the recombination cycle.

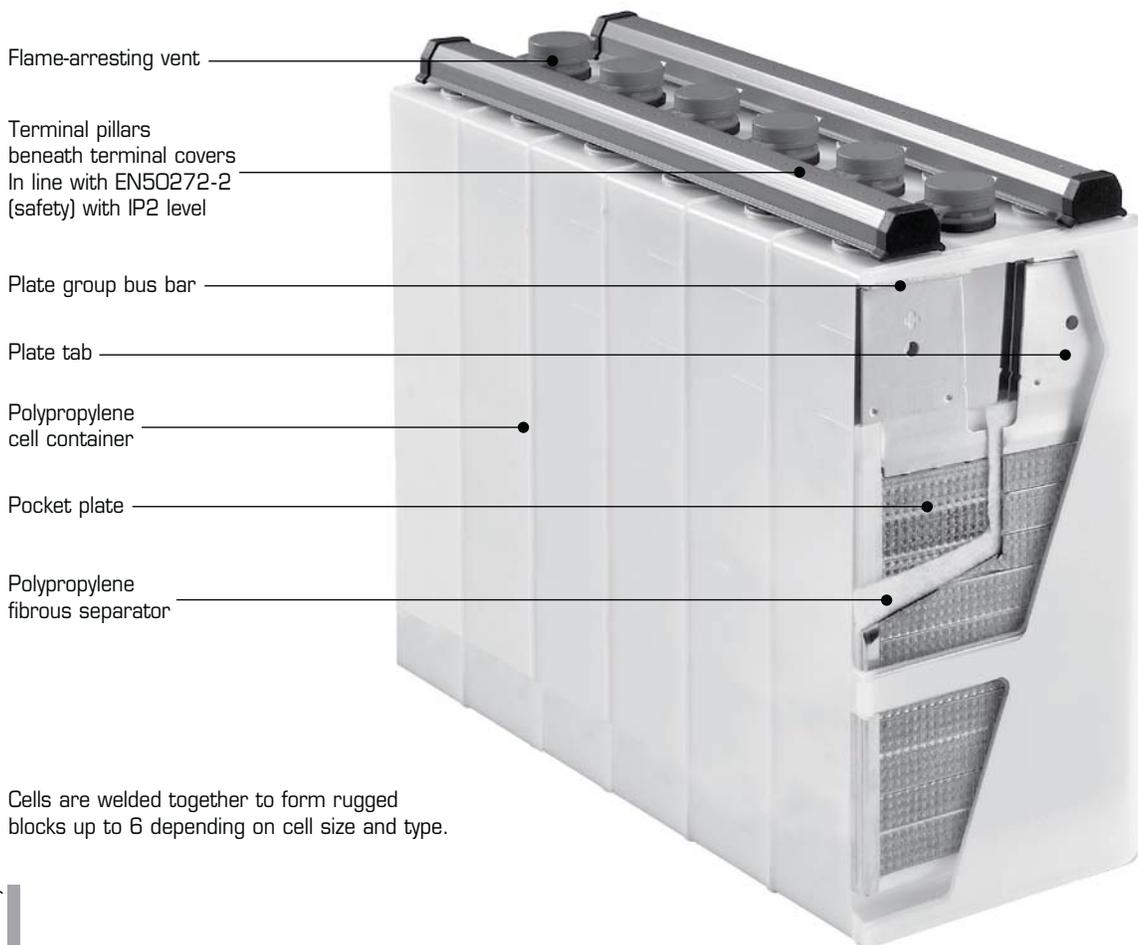
4.1 Plate assembly

The nickel-cadmium cell consists of two groups of plates, one containing nickel hydroxide (the positive plate) and the other containing cadmium hydroxide (the negative plate).

The active materials of the Saft Uptimax pocket plate are retained in pockets formed from nickel-plated steel strips double-perforated by a patented process. These pockets are mechanically linked together, cut to the size corresponding to the plate width and compressed to the final plate dimension. This process leads to a component which is not only mechanically robust but also retains its active material within a steel boundary which promotes conductivity and minimizes electrode swelling.

These plates are then welded to a current carrying bus bar which further ensures the mechanical and electrical stability of the product.

The alkaline electrolyte does not react with steel, which means that the supporting structure of the Uptimax battery stays intact and unchanged for the life of the battery. There is no corrosion and no risk of "sudden death".



4.2 Separation

The separator is a key feature of the Uptimax battery. It is a polypropylene fibrous material which has been used and proven by Saft in the Ultima ultra-low maintenance product over more than 20 years and has been further developed for this product to give the features required. Using this separator, the distance between the plates is carefully controlled to give the necessary gas retention to provide the level of recombination required. By providing a large spacing between the positive and negative plates and a generous quantity of electrolyte between plates, the possibility of thermal runaway, a problem with VRLA cells, is eliminated.

4.3 Electrolyte

The electrolyte used in Uptimax, which is a solution of potassium hydroxide and lithium hydroxide, is optimized to give the best combination of performance, life and energy efficiency over a wide operational temperature range.

The concentration is such as to allow the cell to be operated to temperature extremes as low as -20°C (-4°F) and as high as $+50^{\circ}\text{C}$ ($+122^{\circ}\text{F}$). This allows the very high temperature fluctuations found in certain remote regions to be accommodated. For continuous temperatures below -20°C (-4°F) a special high density electrolyte can be used. It is an important consideration of Uptimax, and indeed all nickel-cadmium batteries, that the electrolyte does not change during charge and discharge. It retains its ability to transfer ions between the cell plates irrespective of the charge level. In most applications the electrolyte will retain its effectiveness for the life of the battery and will never need replacing.

4.4 Terminal pillars

Short terminal pillars are welded to the plate bus bars using a well-proven battery construction method. These posts are manufactured from steel bar, internally threaded for bolting on connectors and are nickel-plated.

The terminal pillar to lid seal is provided by a compressed visco-elastic sealing surface held in place by compression lock washers. This assembly is designed to provide satisfactory sealing throughout the life of the product.

4.5 Venting system

Uptimax is fitted with a flame-arresting flip-top vent to simplify topping-up and is supplied with a transportation plug to ensure safe transportation.

4.6 Cell container

Uptimax is built up using the well-proven Saft block battery construction. The tough polypropylene containers are welded together by heat sealing and the assembly of the blocks is completed by a clip-on terminal cover which gives protection to IP2X according to EN 60529 standard for the conductive parts.

5. Benefits of the Uptimax battery

Complete reliability

Does not suffer from the sudden death failure due to internal corrosion associated with other battery technologies.

Long cycle life

The Uptimax battery has a long cycle life even when the charge/discharge cycle involves 100 % depth of discharge (see section 6.7 Cycling).

Exceptionally long lifetime

A lifetime in excess of twenty years is achieved by Uptimax in many applications, and at elevated temperatures it has a lifetime unthinkable for other widely available battery technologies.

Low maintenance

With its special recombination separator and generous electrolyte reserve, Uptimax reduces the need for topping-up with water. Only one topping-up operation is necessary during its entire service life.

Wide operating temperature range

Uptimax has an electrolyte which allows it to have a normal operating temperature of from -20°C to +40°C (-4°F to +104°F), and accept extreme temperatures, ranging from as low as -40°C to +70°C (-40°F to +158°F) (see section 4.3 Electrolyte).

Fast recharge

Uptimax can be recharged at high currents which allow very fast recharge times to be achieved.

Resistance to mechanical abuse

Uptimax is designed to have the mechanical strength required to withstand all the harsh treatment associated with transportation over difficult terrain (see section 8.2 Mechanical abuse).

High resistance to electrical abuse

Uptimax will survive abuse which would destroy a lead acid battery, for example overcharging, deep discharging, and high ripple currents (see section 8.1 Electrical abuse).

Simple installation

Uptimax can be used with a wide range of stationary and mobile applications as it produces no corrosive vapors, uses corrosion-free polypropylene containers and has a simple bolted connector assembly system (see section 10 Installation and operating instructions).

Well-proven pocket plate construction

Saft has nearly 100 years of manufacturing and application experience with respect to the nickel-cadmium pocket plate

product, and this expertise has been built into the twenty-plus years' design life of the Uptimax product (see section 4 Construction features of the Uptimax battery).

Extended storage

When stored in the filled and charged state in normal condition, Uptimax can be stored for up to 2 years (see section 10 Installation and operating instructions).

Environmentally safe

Saft operates a dedicated recycling center to recover the nickel, cadmium, steel and plastic used in the battery (see section 12 Disposal and recycling).

Low life-cycle cost

When all the factors of lifetime, low maintenance requirements, simple installation and storage and resistance to abuse are taken into account, Uptimax becomes the most cost effective solution for many professional applications.

6. Operating features

6.1 Capacity

The Uptimax battery capacity is rated in ampere-hours (Ah) and is the quantity of electricity at +20°C (+68°F) which it can supply for a 5 hour discharge to 1.0 V/cell after being fully charged. This figure is in agreement with the IEC 62259 standard.

According to the IEC 62259, 0.2 C₅ A is also expressed as 0.2 I_t A. The reference test current (I_t) is expressed as:

$$I_t A = \frac{C_n \text{ Ah}}{1 \text{ h}}$$

where:

C_n is the rated capacity declared by the manufacturer in ampere-hours (Ah), and

n is the time base in hours (h) for which the rated capacity is declared.

In practice, Uptimax is used in floating conditions and so the tabular data is based upon cell performance after several months of floating. This eliminates certain correction factors which need to be used when sizing batteries with conventional fully charged open cell data (see section 9 Battery sizing principles).

6.2 Cell voltage

The cell voltage of nickel-cadmium cells results from the electrochemical potentials of the nickel and the cadmium active materials in the presence of the potassium hydroxide electrolyte. The nominal voltage is 1.2 V.

6.3 Internal resistance

The internal resistance of a cell varies with the type of service and the state of charge and is, therefore, difficult to define and measure accurately.

The most practical value for normal applications is the discharge voltage response to a change in discharge current.

The internal resistance of an Uptimax cell depends on the performance type and at normal temperature has the values given in the product literature for fully charged cells.

For lower states of charge the values increase. For cells 50 % discharged the internal resistance is about 20 % higher, and when 90 % discharged, it is about 80 % higher. The internal resistance of a fully discharged cell has very little meaning.

Reducing the temperature also increases the internal resistance, and at 0°C (+32°F), the internal resistance is about 40 % higher.

The internal resistance of a block battery cell depends on the performance type.

6.4 Effect of temperature on performance

Variations in ambient temperature affect the performance of Uptimax and this needs to be taken into account when sizing the battery.

Low temperature operation has the effect of reducing the performance but the higher temperature characteristics are similar to those at normal temperatures. The effect of temperature is more marked at higher rates of discharge.

The factors which are required in sizing a battery to compensate for temperature variations are given in a graphical form in Figure 1(a) L type and, Figure 1(b) M type for operating temperature -20°C to +40°C (-4°F to +104°F).

6.5 Short-circuit values

The typical short-circuit value in amperes for an Uptimax cell is approximately 6 times the ampere-hour capacity for an L type, 11 times the ampere-hour capacity for an M type. The Uptimax battery is designed to withstand a short-circuit current of this magnitude for many minutes without damage.

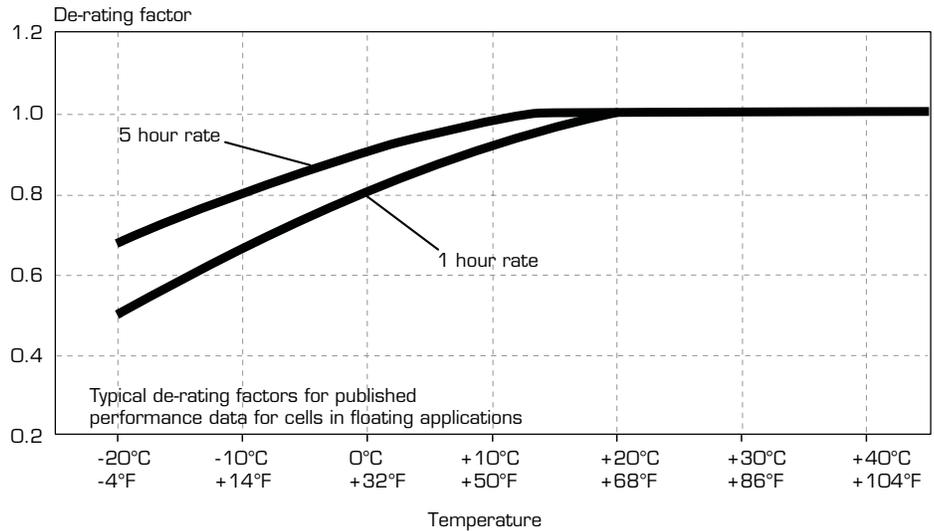


Figure 1(a): Temperature de-rating factors for L type cell for cell final voltage 1.00 V.

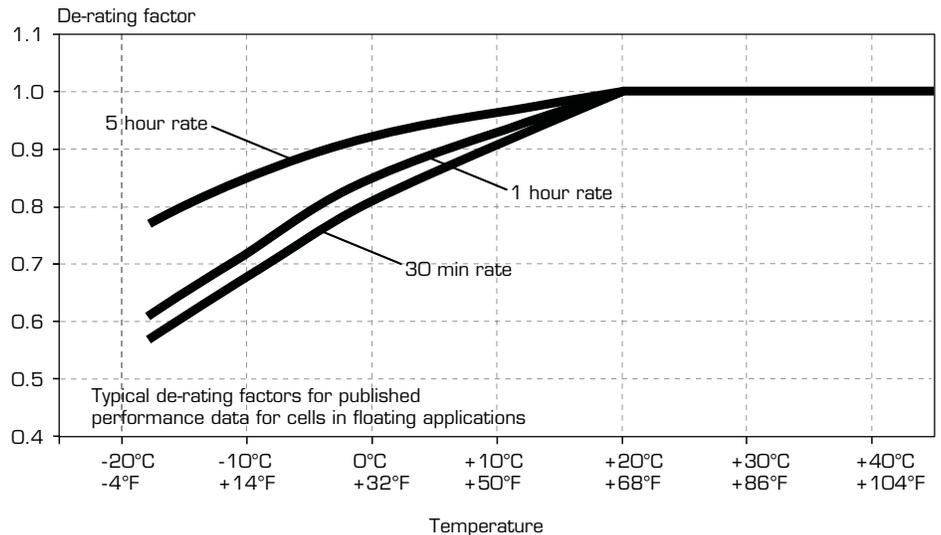


Figure 1(b): Typical de-rating factors for M type cell for cell final voltage 1.00 V.

6.6 Open circuit loss

The state of charge of Uptimax on open circuit slowly decreases with time due to self-discharge. In practice this decrease is relatively rapid during the first two weeks but then stabilizes to about 2 % per month at + 20°C (+ 68°F).

The self-discharge characteristics of a nickel-cadmium cell are affected by the temperature. At low temperatures the charge retention is better than at normal temperature and so the open circuit loss is reduced. However, the self-discharge is significantly increased at higher temperatures.

The open circuit loss for Uptimax for the standard temperature and the extremes of the normal operating range is shown in Figure 2 for a one year period.

6.7 Cycling

Uptimax is a low maintenance product and therefore is used generally in stationary and not continuous cycling applications. Nevertheless, it is designed using conventional pocket plate electrode technology and has therefore an equivalent cycling capability to the standard product.

If Uptimax is used in a deep cycling application which requires a fast recharge, there will be significant gas evolved and the low maintenance properties of the product will be severely reduced. However, there are cycling applications where Uptimax can be beneficial. This will depend on the frequency and depth of discharge involved.

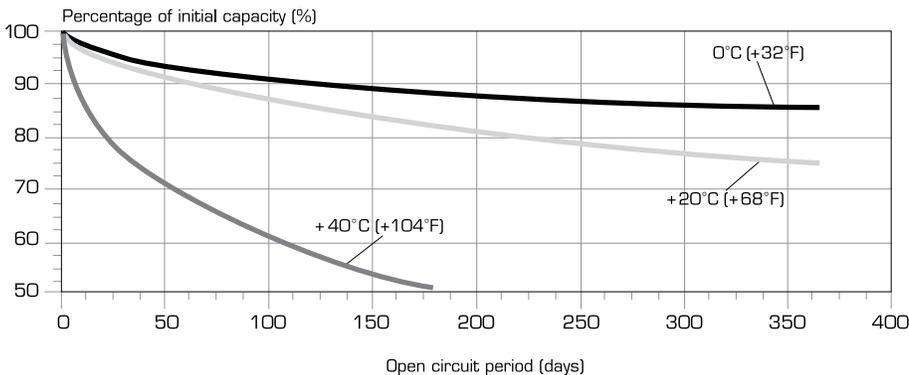


Figure 2: Typical open circuit loss variation with time.

6.8 Water consumption

The Uptimax battery works on the oxygen recombination principle and therefore has a much reduced water consumption. In practice, for the recommended charging voltages, Uptimax has a level of recombination of more than 90 %. This compares to the level of recombination found in equivalent vented pocket plate cells of 30 % to 35 %. Thus Uptimax has a water usage reduced by a factor of up to 10 times of that of an open flooded cell. This means that at suitable charging voltages, Uptimax will need only one topping-up operation during its entire service life.

However, not all needs are the same for water replenishment under different and more difficult charging conditions.

6.9 Gas evolution

The gas evolution is a function of the amount of water electrolyzed into hydrogen and oxygen which is not involved in the recombination cycle. The electrolysis of 1 cm³ of water produces about 1865 cm³ of gas mixture and this gas mixture is in the proportion of 2/3 hydrogen and 1/3 oxygen. Thus the electrolysis of 1 cm³ of water produces about 1243 cm³ of hydrogen.

As stated in section 6.8, under normal recommended float conditions Uptimax has a recombination level of more than 90 % and so the amount of water which is electrolyzed into gas is small. Typically an Uptimax cell will electrolyze about 0.002 cm³ of water per Ah of cell capacity per day. This value will be smaller or larger depending on the float voltage value. Thus a typical value of gas emission would be 4.5 cm³ per Ah of cell capacity per day, or 3 cm³ of hydrogen per Ah of cell capacity per day.

7. Battery charging

In order to ensure that the low maintenance properties of the Uptimax battery are achieved, it is necessary to control the charge input to the battery to minimize the rate of water loss during the life of the product.

It is important therefore that the recommended charge conditions are complied with. However, Uptimax has a high technology, low maintenance concept in allowing the possibility of replenishment of water in severe applications where excessive water loss is unavoidable.

7.1 Charging methods

Uptimax batteries may be charged quickly and simply by the following methods:

a) Two level constant potential charging

The initial stage of two-rate constant potential charging consists of a first charging stage to a maximum voltage of 1.45 ± 0.01 V/cell.

Alternatively, if a faster rate of recharge is required, a voltage limit of 1.55 V/cell with a current limit of 0.1 C₅A can be used. However, if frequent recharges are required this will increase the rate of water loss and gas generation.

After this first stage the charger should be switched to a second maintenance stage at a float voltage of 1.43 ± 0.01 V/cell. After a prolonged mains failure the first stage should be reapplied manually or automatically.

b) Single level float charging

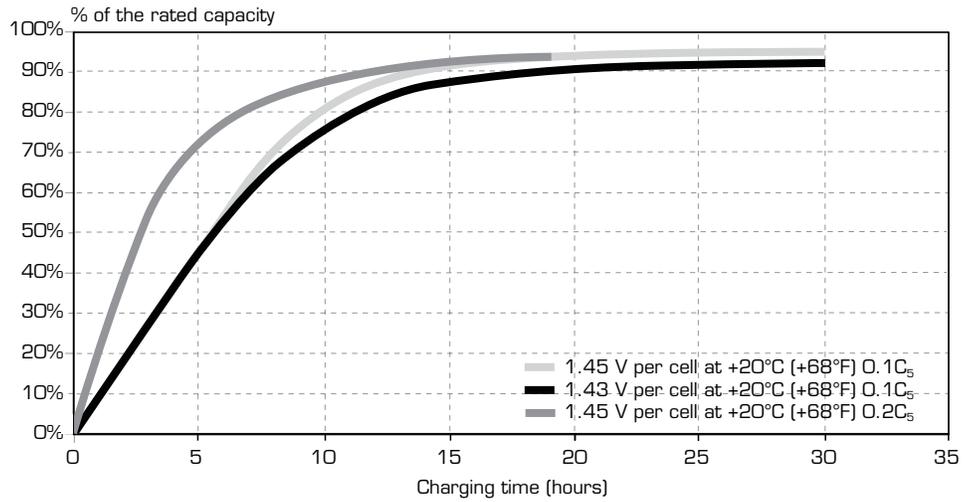
Uptimax batteries are float charged at 1.43 ± 0.01 V/cell from a fully discharged condition to a high state of charge. This is detailed in section 7.2 and about 90 % of the capacity will be available after 15 hours of charge even at +40°C (+104°F).

7.2 Charge acceptance

The performance data sheets for Optimax are based upon several months' floating and so are for fully float charged cells.

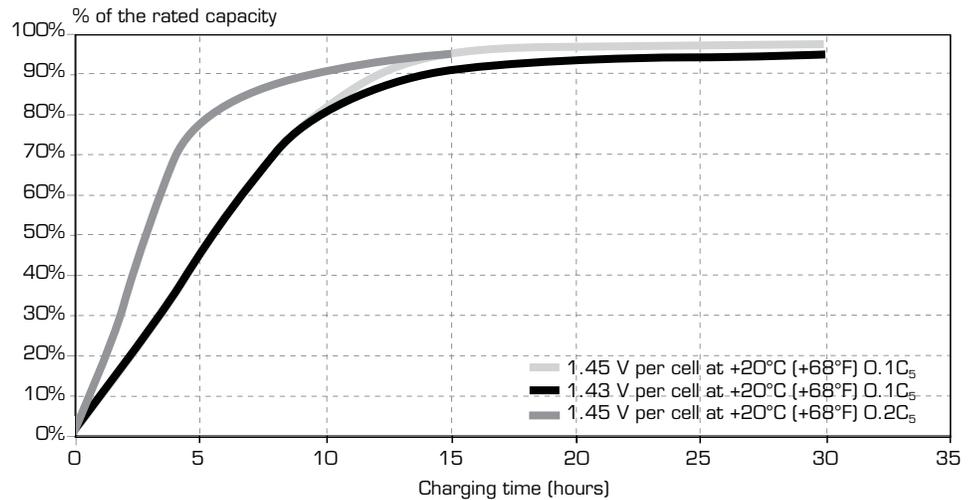
A discharged cell will take a certain time to achieve this and Figure 3 gives the capacity available for the two principal charging voltages recommended for Optimax, 1.43 V/cell and 1.45 V/cell, during the first 30 hours of charge from a fully discharged state.

If the application has a particular recharge time requirement then this must be taken into account when calculating the battery.



**For charging voltages higher than 1.45 V/cell, a current limit of 0.1 C₅ is recommended*

Figure 3(a): Available capacity after constant voltage charge Available charge current 0.1C₅ A or 0.2C₅ A, for L type cell.



**For charging voltages higher than 1.45 V/cell, a current limit of 0.1 C₅ is recommended*

Figure 3(b): Available capacity after constant voltage charge Available charge current 0.1C₅ A or 0.2C₅ A, for M type cell.

7.3 Charge efficiency

The charge efficiency of Uptimax is dependent on the state of charge of the battery and the temperature. For much of its charge profile it is recharged at a high level of efficiency. In general, at states of charge less than 80 % the charge efficiency remains high, but as the battery approaches a fully charged condition, the charging efficiency falls off.

7.4 Temperature effects

As the temperature increases, then the electrochemical behaviour becomes more active and so, for the same charge voltage, the current at the end of charge increases. This end of charge increases in the current helps to compensate for the variation in charge efficiency at high temperatures and allows a high state of charge to be achieved. For this reason it is **not needed to apply** temperature compensation of the charge voltage used for ambient temperatures above 10°C (50°F). To minimize water consumption at sustained operation at high temperatures, temperature compensation can be used, to reduce the increase of current at higher temperatures.

As the temperature is reduced, then the reverse occurs and it is recommended that, for application where the ambient temperature falls below 10°C (50°F) for sustained periods, temperature compensation of the charge voltage should be used to maintain the end of charge current at a constant value.

When temperature compensation is used the

change in voltage required per cell, or "temperature compensation", should be between -2 mV and -3.5 mV per °C (-1.12 mV and -1.96 mV per °F). The recommended value per cell is -2.5 mV per °C (-1.4 mV per °F).

If main operation is at high temperature, **use** -2 mV per °C (-1.12 mV per °F).

If the fast recharge at low temperature (below -10°C (14°F)) is very important, **use** -3.5 mV per °C (-1.96 mV per °F).

8. Special operating factors

8.1 Electrical abuse

8.1.1 Ripple effects

The nickel-cadmium battery is tolerant to high ripple from standard charging systems. Uptimax accepts ripple currents up to $0.2 C_5 A I_{eff}$. In general, any commercially available charger or generator can be used for commissioning or maintenance charging of Uptimax.

8.1.2 Over-discharge

If more than the designed capacity is taken out of a battery then it becomes over-discharged. This is considered to be an abuse situation for a battery and should be avoided.

In the case of lead acid batteries this will lead to failure of the battery and is unacceptable.

The Uptimax battery is designed to make recovery from this situation possible.

8.1.3 Overcharge

Overcharge is the effect of forcing current through a battery when it is fully charged. This can be damaging for a lead acid battery, and due to its starved electrolyte technology, seriously reduce the life of a VRLA battery.

In the case of Uptimax, with its generous electrolyte reserve, a small degree of overcharge will not significantly alter the maintenance period. In the case of excessive overcharge, water replenishment is required but there will be no significant effect on the life of the battery.

8.2 Mechanical abuse

8.2.1 Shock loads

The Uptimax block battery concept has been tested to IEC 68-2-29 (bump tests at 5 g, 10 g and 25 g) and IEC 77 (shock test 3 g), where g = acceleration.

8.2.2 Vibration resistance

The Uptimax block battery concept has been tested to IEC 77 for 2 hours at 1 g, where g = acceleration.

8.2.3 External corrosion

Uptimax nickel-cadmium cells are manufactured in durable polypropylene, all external metal components are nickel-plated and these components are protected by an anti-corrosion oil and a rigid plastic cover.

9. Battery sizing principles in stationary applications

There are a number of methods which are used to size nickel-cadmium batteries for standby floating applications. The method employed by Saft is the IEEE 1115 recommendation which is accepted internationally. This method takes into account multiple discharges, temperature de-rating, performance after floating and the voltage window available for the battery.

A significant advantage of the nickel-cadmium battery compared to a lead acid battery, is that it can be fully discharged without any inconvenience in terms of life or recharge. Thus, to obtain the smallest and least costly battery, it is an advantage to discharge the battery to the lowest practical value in order to obtain the maximum energy from the battery.

The principle sizing parameters which are of interest are:

9.1 The voltage window

This is the maximum voltage and the minimum voltage at the battery terminals acceptable for the system. In battery terms, the maximum voltage gives the voltage which is available to charge the battery, and the minimum voltage gives the lowest voltage acceptable to the system to which the battery can be discharged. In discharging the nickel-cadmium battery, the cell voltage should be taken as low as possible in order to find the most economic and efficient battery.

9.2 Discharge profile

This is the electrical performance required from the battery for the application. It may be expressed in terms of amperes for a certain duration, or it may be expressed in terms of power, in watts or kW, for a certain duration. The requirement may be simply one discharge or many discharges of a complex nature.

9.3 Temperature

The maximum and minimum temperatures and the normal ambient temperature will have an influence on the sizing of the battery. The performance of a battery decreases with decreasing temperature and sizing at a low temperature increases the battery size. Temperature de-rating curves are produced for all cell types to allow the performance to be recalculated.

9.4 State of charge or recharge time

Some applications may require that the battery shall give a full duty cycle after a certain time after the previous discharge. The factors used for this will depend on the depth of discharge, the rate of discharge, and the charge voltage and current. A requirement for a high state of charge does not justify a high charge voltage if the result is a high end of discharge voltage.

9.5 Ageing

Some customers require a value to be added to allow for the ageing of the battery over its lifetime. This may be a value required by the customer, for example 10 %, or it may be a requirement from the customer that a value is used which will ensure the service of the battery during its lifetime. The value to be used will depend on the discharge rate of the battery and the conditions under which the discharge is carried out.

9.6 Floating effect

When a nickel-cadmium cell is maintained at a fixed floating voltage over a period of time, there is a decrease in the voltage level of the discharge curve. This effect begins after one week and reaches its maximum in about 3 months. It can only be eliminated by a full discharge/charge cycle, and it cannot be eliminated by a boost charge. It is therefore necessary to take this into account in any calculations concerning batteries in float applications.

This is used in the IEEE sizing method and the published data for Uptimax.

As the effect of reducing the voltage level is to reduce the autonomy of the battery, the effect can be considered as reducing the performance of the battery and so performance down-rating factors are used.

10. Installation and operating instructions

Type UP1 L and UP1 M

Important recommendations

- Never allow an exposed flame or spark near the batteries, particularly while charging.
- Never smoke while performing any operation on the battery.
- For protection, wear rubber gloves, long sleeves, and appropriate splash goggles or face shield.
- The electrolyte is harmful to skin and eyes. In the event of contact with skin or eyes, wash immediately with plenty of water. If eyes are affected, flush with water, and obtain immediate medical attention.
- Remove all rings, watches and other items with metal parts before working on the battery.
- Use insulated tools.
- Avoid static electricity and take measures for protection against electric shocks.
- Discharge any possible static electricity from clothing and/or tools by touching an earth-connected part “ground” before working on the battery.

10.1 Receiving the shipment

Unpack the cells immediately upon arrival. Do not overturn the package. Check the packages and cells for transport damage.

The cells are shipped filled and charged, and are ready for immediate use.

Transport seals are located under the lid of each vent and they must be removed prior to charging.

The cells must never be charged with the plastic transport seals in place as this is dangerous and can cause permanent damage.

10.2 Storage

Store the battery indoors in a dry, clean, cool location (0°C to +30°C / +32°F to +86°F) and well ventilated space on open shelves.

Storage of a filled battery at temperatures above +30°C (+86°F) can result in loss of capacity. This can be as much as 5 % per 10°C (18°F) above +30°C (+86°F) per year.

Do not store in direct sunlight or expose to excessive heat.

Uptimax cells are supplied filled with electrolyte and charged, **they can be stored in this condition for maximum 24 months.**

Never drain the electrolyte from the cells.

- When deliveries are made in cardboard boxes, store without opening the boxes.
- When deliveries are made in plywood boxes, open the boxes before the storage. The lid and the packing material on top of the cells must be removed.

10.3 Installation

10.3.1 Location

Install the battery in a dry and clean room. Avoid direct sunlight and heat. The battery will give the best performance and maximum service life when the ambient temperature is between +10°C to +30°C (+50°F to +86°F).

10.3.2 Ventilation

During the last part of charging, the battery is emitting gases (oxygen and hydrogen mixture). At normal float charge the gas evolution is very small but some ventilation is necessary.

Note that special regulations for ventilation may be valid in your area depending on the application.

10.3.3. Mounting

Verify that cells are correctly interconnected with the appropriate polarity. The battery connection to load should be with nickel-plated cable lugs. Apply a thin layer of anti-corrosion oil to protect the connectors and terminals from corrosion.

Recommended torques for terminal bolts are:

- M6 = 11 ± 1.1 N.m
(97.4 ± 9.7 lbf.in)
- M8 = 20 ± 2 N.m
(177.0 ± 17.7 lbf.in)
- M10 = 30 ± 3 N.m
(265.5 ± 26.6 lbf.in)

The connectors and terminal should be corrosion-protected by coating with a thin layer of anti-corrosion oil.

Remove the transport seals and close the vent plugs.

10.3.4 Electrolyte

When checking the electrolyte levels, a fluctuation in level between cells is not abnormal and is due to the different amounts of gas held in the separators of each cell. The level should be at least 15 mm above the minimum level mark (lower) and there is normally no need to adjust it.

In case of spillage of electrolyte during the transport, the cells have to be topped up with E22 electrolyte.

Do not top-up prior to initial charge.

10.4 Commissioning

Verify that the transport seals are removed, the vents are closed, and the ventilation is adequate during this operation.

A good commissioning is important.

■ Cells stored up to 6 months:

A commissioning charge is normally not required and the cells are ready for immediate use.

■ Cells stored more than 6 months and up to 2 years:

A commissioning charge is necessary:

• Constant current charge:

Charge for 16 h at $0.1 C_5$ A recommended (see Installation and operating instructions sheet).

Note: At the end of charge, the cell voltage may reach the level of 1.85 V, thus the charger shall be able to supply such a voltage.

When the charger maximum voltage setting is too low to supply constant current charging, divide the battery into two parts to be charged individually at constant current.

- **Constant voltage charge:**

Charge for 1.65 V/cell for a minimum of 30 h with current limited to 0.1 C₅ A (see Installation and operating instructions sheet).

If these methods are not available, then charging may be carried out at lower voltages, 1.50 V/cell for 72 hours minimum.

Note: For capacity test purposes, the battery has to be charged in accordance with IEC 62259 section 7 (7.1 & 7.2).

10.5 Charging in service

The recommended charging voltages for continuous parallel operation, with occasional battery discharges, are:

- **Two level charge:**

- float level:

1.43 ± 0.01 V/cell

- high rate (boost) level:

1.45 ± 0.01 V/cell

- **Single level charge:**

1.43 ± 0.01 V/cell

10.6 Periodic maintenance

Uptimax is a low maintenance battery and requires the minimum of maintenance.

As a periodic maintenance, the following is recommended:

- Keep the battery clean using only water. Do not use a wire brush or solvents of any kind.
- Check visually the electrolyte level.

Never let the level fall below the minimum level mark (lower). Use only distilled or de-ionized water to top-up.

Topping-up of the Uptimax battery **shall be carried out when battery is fully charged.**

Experience will tell the time interval between topping-up.

Note: *There is no need to check the electrolyte density periodically. Interpretation of density measurements is difficult and could be misleading.*

- Check the charging voltage.

In parallel operation, it is of great importance that the recommended charging voltage remains unchanged. The charging voltage should be checked and recorded at least once yearly. If a cell float voltage is found below 1.35 V, high-rate charge recommended to apply to the cell concerned.

- Check every two years that all connectors are tight. The connectors and terminal bolts should be corrosion-protected by coating with a thin layer of anti-corrosion oil.
- High water consumption of the battery is usually caused by improper voltage setting of the charger.

11. Maintenance of Uptimax batteries in service

In a correctly designed stationary application, Uptimax requires the minimum of attention.

However, it is good practice with any system to carry out an inspection of the system once per year or at the recommended topping-up interval period to ensure that the charging system, the battery and the ancillary electronics are all functioning correctly.

When this system service is carried out, it is recommended that the following actions should be taken:

- Cell electrolyte levels should be checked visually to ensure that the level is above the minimum and if necessary the cells should be topped-up. Use only distilled or deionized water.
- The batteries should also be checked for external cleanliness, and if necessary cleaned with a damp brush using water. Do not use a wire brush or solvents of any kind. Vent plugs can be rinsed in clean water if necessary.
- All the connectors must be tight. The connectors and terminal bolts should be corrosion-protected by coating with a thin layer of anti-corrosion oil.

12. Disposal and recycling

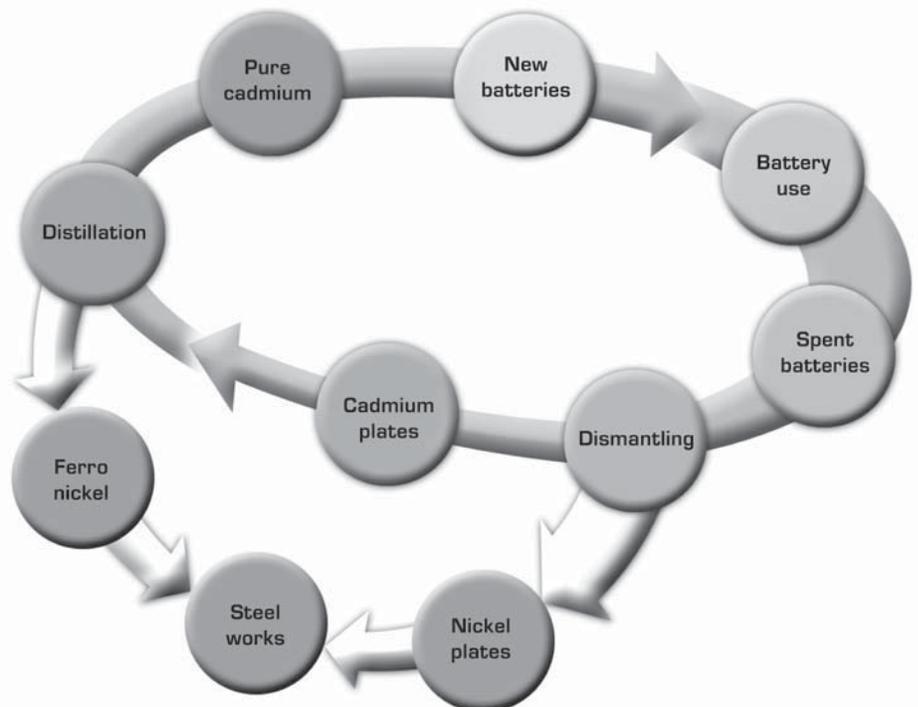
In a world where autonomous sources of electric power are ever more in demand, Saft batteries provide an environmentally responsible answer to these needs. Environmental management lies at the core of Saft's business and we take care to control every stage of a battery's life cycle in terms of potential impact. Environmental protection is our top priority, from design and production through end-of-life collection, disposal and recycling.

Our respect for the environment is complemented by an equal respect for our customers. We aim to generate confidence in our products, not only from a functional standpoint, but also in terms of the environmental safeguards that are built into their life cycle. The simple and unique nature of the battery components make them readily recyclable and this process safeguards valuable natural resources for future generations.

In partnership with collection agencies worldwide, Saft organizes retrieval from pre-collection points and the recycling of spent Saft batteries. Saft's collection network can be found on our web site:

www.saftbatteries.com

Ni-Cd batteries must not be discarded as harmless waste and should be treated carefully in accordance with local and national regulations. Your Saft representative can assist with further information on these regulations and with the overall recycling procedure.





Soft is committed to the highest standards of environmental stewardship.

As part of its environmental commitment, Soft gives priority to recycled raw materials over virgin raw materials, reduces its plants' releases to air and water year after year, minimizes water usage, reduces fossil energy consumption and associated CO₂ emissions, and ensures that its customers have recycling solutions available for their spent batteries.

Regarding industrial Ni-Cd batteries, Soft has had partnerships for many years with collection companies in most EU countries, in North America and in other countries. This collection network receives and dispatches our customers' batteries at the end of their lives to fully approved recycling facilities, in compliance with the laws governing trans-boundary waste shipments. This collection network is undergoing minor adaptations to meet the requirements of the EU batteries directive. A list of our collection points is available on our web site.

In other countries, Soft assists users of its batteries in finding environmentally sound recycling solutions. Please contact your sales representative for further information.

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